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## CHEMICAL AND MICROBIAL CHANGES DURING THE NATURAL FERMENTATION OF ARBUTUS UNEDO'S FRUITS

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Complete List of Authors:	Carvalho, Isabel; University of Algarve, Biotechnology Cavaco, Teresa; University of Algarve, Biotechnology Quintas, Célia; University of Algarve, EST Longuinho, Carla; University of Algarve, Biotechnology
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27 **Key words: fruit from *Arbutus unedo*, natural fermentation, sugars, mineral**  
28 **composition**

## INTRODUCTION

31 *Arbutus unedo* L. (strawberry tree) (family: *Ericaceae*) is an evergreen shrub  
32 usually smaller than 4 m found in the Mediterranean fringe, being today also cultivated  
33 in the Near East and Transcaucasia (Seidemann, 1995). The fruits of strawberry tree are  
34 spherical berries of about 1-2 cm in diameter, dark red when mature in autumn. They  
35 are tasty and processed into jam, fruit jelly, spirits and liqueurs being rarely eaten as  
36 fresh fruit. The fruits and leaves are also well known in folk medicine as antiseptics,  
37 diuretics and laxatives in Turkey (Ayat *et al.*, 2000; Pabuçcuoglu *et al.*, 2003).

38 In Portugal strawberry tree has a remarkable implantation in the south (Algarve)  
39 where it has been associated with a long tradition in social, cultural and microeconomic  
40 aspects. In the Algarve's region the fruits, when fully ripe, are fermented and used in the  
41 manufacture of a "traditional product", an alcoholic distillate named Aguardente de  
42 medronho (medronho firewater), very appreciated locally for its flavor characteristics.  
43 This spirit has been a complement to farmers' income, for generations, and over the last  
44 years efforts have been made to improve the quality of the production and the quality of  
45 the final product. The preservation and valorization of traditional resources are of great  
46 importance to prevent their disappearance (Alarcão e Silva *et al.*, 2001; Soufleros *et al.*,  
47 2005).

48 Fermentations are done on the farm and the results are very variable in quality.  
49 There are problems of acidity or lack of flavor or even off-flavors due to uncontrolled  
50 fermentations (incomplete fermentations or over fermentations) all of which leading to

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51 low crop value for the farmer. It would be desirable to study the process in order to  
52 improve the quality of the final product (medronho firewater).

53 In the Algarve's region, about ten thousand liters of this distilled are produced per  
54 year, on a small scale as a home made product. Medronho firewater is made from  
55 recently harvested mature fruits mixed with water in closed recipients, where alcoholic  
56 fermentation occurs due to wild microbiota, at room temperature, during 4-5 weeks,  
57 depending on the weather conditions (temperature). After the fermentation the  
58 fermented material is heated by direct fire in a copper still and the condensation of both  
59 ethanol and non-ethanolic volatiles are crucial for the flavor and aroma of the final  
60 product. The distillate obtained has 40 to 50 % ethanol (v/v) and can be matured in  
61 wooden barrels for several years. In general, the fermentation and distillation processes  
62 are done between September and January.

63 The nature of the fermentation with respect to the role played by the  
64 microorganisms and their enzymes on the physicochemical changes and on the  
65 characteristics of the final product is not known. Therefore, the aim of this work is to  
66 study the microbial population dynamics, the changes in sugar composition and some  
67 fermentation by-products as ways of monitoring the fermentation evolution.

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## 69 MATERIALS AND METHODS

### 70 Fruit Sampling

71 Strawberry tree (*Arbutus unedo* L.) fruits used in this study were obtained from  
72 trees grown in Caldeirão Saw, Algarve (Portugal), in their normal harvest time.  
73 Randomly, three pickings, of fruit at red mature stage (12-23 °Brix), were made  
74 between September and November 2004. The fruits were transported to the farmers in  
75 polyethylene bags and kept away from direct sunlight to avoid qualitative losses. Two,

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3 76 200Kg, samples of fruit were used for the fermentation process. Three, 200g, samples of  
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5 77 ground sample were used for the extractions of sugars, ethanol, organic acids and  
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8 78 minerals. All determinations were made from these extracts in triplicate. For chemical  
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10 79 purposes, the samples were kept in freeze (-20°C) until analyzed.  
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#### 12 **Fermentation conditions**

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15 81 Mature fruits recently harvested, were mixed with water and partially crushed (1  
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17 82 part of water to 10 parts of fruit). Fermentation took place due to indigenous microbiota  
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19 83 in closed PVC recipients (200Kg-250Kg), at room temperature, during 36 days.  
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21 84 Samples were taken every 3 days during 36 days from the fermentation recipient. Each  
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23 85 sample consisted of about 100g of fermented must removed with sterile tongs from  
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25 86 different points approximately 30cm from walls and 45cm from upper surface. The  
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27 87 samples were aseptically transferred in PVC bags to the laboratory, under refrigerating  
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29 88 conditions, and analyzed, for microbial counts in the same day. For chemical purposes  
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31 89 samples were kept in freeze (-20°C) until analyzed.  
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#### 36 **Determination of pH and total amount of Titratable Acids (TTA)**

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38 91 The pH and the TTA were determined in every sample with a Delta 320  
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40 92 pH meter (Mettler, France). A sample (10g) was mixed with distilled water (90mL) and  
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42 93 the mixture was then totated with NaOH (0.1 M) with stirring to pH 8.5. The total  
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44 94 titratable acidity was expressed as the amount of Na OH consumed, in milliliters.  
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#### 48 **Sugars and Ethanol**

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50 96 The mature samples (15g) were homogenized at room temperature, clarified by  
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52 97 centrifuging at 4000 rpm (21°C) for 20 min, and filtered through Sep- Pak C18 (Waters  
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54 98 Associates) according to Medlicott and Thompson (1985). The sugars and ethanol  
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56 99 content were analyzed by HPLC with a Beckman System Gold Chromatograph  
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58 100 (Fullerton, U.S.A.) composed of a 126 gradient pump with a detector UV-Vis Beckman  
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3 101 166 (USA). Sample injection was by means of rheodyne injection valve (Rheodyne,  
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5 102 California, USA) equipped with polyspher (OA HY) column the mobile phase was  
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7 103 acetonitrile and water (20:80) at a flow rate of 1.0 mL.min<sup>-1</sup>. Individual sugars and  
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9 104 ethanol were quantified by comparison of peak areas of samples to those of standard  
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11 105 curves of ternary mixture solutions with concentrations of 5 mg.mL<sup>-1</sup> (sucrose, glucose  
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13 106 and fructose) available from Sigma Chemical Co, (St. Louis, MO, U.S.A. 2004) and  
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15 107 dried for 12 h at 60°C under vacuum before being dissolved in water. The standard  
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17 108 solutions were stored at -18°C.

### 109 **Organic acids**

110 The same extract obtained for sugars and ethanol analysis was used in the  
111 quantification of organic acids. The identification of organic acids was performed by  
112 HPLC with a detector UV-Vis Beckman 166 (USA). The eluates were detected at 214  
113 nm, on a µBomdapak-C<sub>18</sub> column (waters Associates) with 0.2M KH<sub>2</sub>PO<sub>4</sub> adjusted to  
114 pH 2.4 (H<sub>3</sub>PO<sub>4</sub>) as mobile phase and 1.0 mL.min<sup>-1</sup>.

### 115 **Mineral Analysis**

116 From each homogenized sample 20g were taken in duplicate for the study of  
117 mineral composition. These underwent drying in a thermoregulable swindle at 90°±50°C  
118 overnight. Removal of the organic matter was carried out by means of incineration in a  
119 muffle oven at 450°± 250°C until ash was obtained. Then the ash was dissolved in 2ml  
120 of HCL 1:1 (v/v) maintaining it in an ultrasound bath until complete dissolution, later  
121 removing the acid by evaporation on a thermoregulable sand bath. The residue was  
122 dissolved in deionised water and adjusted to a volume of 50ml and 100ml, for the  
123 determinations of micro and macroelements, respectively. The determinations were  
124 carried out on the solutions although, in the case of macroelements, it was necessary to  
125 dilute them by 1:10. Conditions were carefully controlled to avoid contamination or loss

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3 126 of materials. Ca, K and Na were determined by flame photometry. Intra-assay (n=10)  
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5 127 coefficients of variation (CV) were 3.4, and 2.0% for K and Na, respectively. Recovery  
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7 128 of added known amounts of standards to samples gave 98-105% of expected values for  
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9 129 all minerals. The elements Mg, Fe, Zn, Mn, Pb, Cd and Cu, were measured by atomic  
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11 130 absorption spectrophotometer, equipped with hollow cathode lamps. Coefficient of  
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13 131 variation of the overall method was 7.7%. Results are the mean of triplicate  
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15 132 determinations and expressed as g/ 100g dry matter.  
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### 19 **Microbiological parameters**

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22 134 Twenty five gram samples of the must in fermentation, obtained as described  
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24 135 above at different time intervals, were mixed with 225 ml of sterile 0.1 % peptone water  
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26 136 in a stomacher, during 3 minutes. Sequential decimal dilutions were made with sterile  
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28 137 0.1 % peptone water and aliquots were plated, by surface spreading, in duplicate on the  
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30 138 following agar media for the detection and enumeration of microorganisms. The media  
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32 139 used was: a) Total viable count: Plate count agar (PCA) (pH 4) incubated at 25°C for 5  
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34 140 days; b) Lactic acid bacteria: Man, Rogosa and Sharpe agar (MRS) to which  
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36 141 cycloheximide (0,05%) was added to inhibit growth of contaminant yeasts and molds,  
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38 142 incubated at 25°C for 5 days; c) Acetic acid bacteria: medium containing yeast extract  
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40 143 (10 g), CaCO<sub>3</sub> (20 g), ethanol (20 ml), agar (20 g) to a final volume of 1000 ml; d)  
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42 144 Yeasts: Malt Extract Agar (MEA) (pH 3,6) incubated at 25°C for 5 days. All the  
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44 145 microorganisms were enumerated and the yeasts were isolated and frozen (-80°C) for  
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46 146 identification in future work.  
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### 51 **Statistical analysis**

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54 148 Significant differences between means of experiments were determined by least  
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56 149 significant difference. A significance level of 0.05 was chosen (Sokal and Rohlf, 1987).  
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## RESULTS AND DISCUSSION

Fermentation of *Arbutus unedo* fruits took place on the farm at room temperature. External minimum and maximum temperatures and the fermentative recipient temperature are shown in Table I. The maximum external temperature registered was 18,0°C and the minimum was 9,0°C. On the other hand, the maximum temperature in the fermentation vessel was 15,9°C and the minimum was 12,4°C, in the last day of fermentation.

The changes in hydrogen ion concentration (pH) and °Brix during the fermentation process are reported in Table 2. The pH increased slightly and °Brix decreased during the fermentative process probably due to the production of ethanol resulting from consumption of sugars by fermenting microorganisms.

The mineral composition is presented in Table 3. From the results obtained during fermentation period, the amounts of Fe, Zn, Mn, Cu, Pb and Cd (data not show) were underneath the quantification limit for detection. The importance of the Cu, Pb and Cd analysis is due to their toxicity, that most derives from the alembics (Soufleros *et al.*, 2005). During fermentation period, the mineral content (expressed as mg.100g<sup>-1</sup> dry matter), was increasing resulting in a final product enriched in Ca, Mg, Na and K, where K and Ca are the main elements. At the same time, once phytic acid is well known by chelating and precipitating multivalent cationic minerals with its six anionic phosphate groups, this increase could be explained by the presence of phytate- degrading enzymes present in yeast (Turk *et al.*, 2000) which could release those minerals.

The content of sugars and acids in fruits influences markedly their sensory quality and consequently the quality of the final product. For these reasons, fruit compositional analysis is of interest to food experts. The soluble sugars identified and quantified in the strawberry tree's fruits and all over the fermentation period were fructose, glucose and

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3 176 sucrose (Fig. 1). The first two sugars were the major sugars while sucrose was in much  
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6 177 lesser amounts. One liter of broth of fermentation in the beginning of the process (day  
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8 178 0) contained about 7.3% of fructose, 2.5% of glucose and 4.5% of sucrose at the end of  
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10 179 fermentation (day 36) one liter of broth of fermentation contained 2.5% of fructose,  
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12 180 0.2% of glucose and 0.1% of sucrose. Glucose and fructose are generally considered the  
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14 181 main sugars in strawberry tree's fruits (Hegnauer, 1986; Ayaz *et al.*, 2000). According  
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16 182 to Ayaz *et al.*, (2000) in these types of fruits, the content of sucrose is lower than the  
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18 183 amount of fructose and glucose. The explanation may be that the sucrose, synthesized in  
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20 184 the leaves, is enzymatically hydrolyzed to glucose and fructose when translocated to the  
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22 185 fruit.  
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27 186 During the firsts two days of the fermentation period the levels of fructose and  
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29 187 glucose increased reaching their maximum (glucose-66.8 g.L<sup>-1</sup>; fructose-127.978g.L<sup>-1</sup>).  
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31 188 This increasing phase was followed by a decline in their contents in the fermentative  
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33 189 must accompanied by a rising in the concentration of ethanol that varied from 1.47×10<sup>-2</sup>  
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35 190 l per liter l of fermentation broth in the beginning of the study (day 2) to 7.6×10<sup>-2</sup> l per  
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37 191 liter of fermentation broth in the end of the experiment (day 36).  
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41 192 Fig. 1 shows the evolution of soluble sugars (glucose, fructose and sucrose) and  
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43 193 yeast population during fermentation of *Arbutus unedo* fruits. The variations in sugars  
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45 194 and ethanol contents are related with the activities of the microbial population. In  
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47 195 addition, the presence of vestigial concentrations of organic acids is probably related  
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49 196 with the inexistence of lactic or acetic acid bacteria.  
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53 197 All over the fermentation period, the number of lactic acid bacteria and acetic acid  
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55 198 bacteria detected was inferior to 10 cfu/ml of must and the total viable counts were  
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57 199 approximately equal or even less then the yeasts counts. The results presented in this  
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59 200 study show that, yeasts were the main responsible for the fermentation of the fruits of  
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3 201 *Arbutus unedo* in the process monitored. The colonization and subsequent growing of  
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5 202 these microorganisms were favored by the initial acidity of the pulp' fruits (pH 3.8).  
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8 203 The yeast population in the fermentation must was initially high ( $1.25 \times 10^5$  cfu/ml of  
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10 204 must). As can be seen in Fig 1, during the firsts 8 days yeasts showed an exponential  
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12 205 phase that lasted about 8 days, reaching a number of  $2 \times 10^7$  cfu/ml of must. Next, a  
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14 206 stationary phase occurred almost all over the period studied. The growing of the yeasts  
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16 207 was accompanied by a decreasing of sugars contents in the must and their conversion to  
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18 208 ethanol which is a primary activity of yeasts during alcoholic fermentation. The content  
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20 209 of glucose and fructose reached their highest value on the second day of fermentation.  
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22 210 The initial crushing and mixture with water caused the rupture of the fruits' skin,  
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24 211 facilitating the contact between the fruits' flesh and the microorganisms. During this  
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26 212 period the indigenous flora produce hydrolytic enzymes which break down cell  
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28 213 components and complex carbohydrates, producing simple molecules, including  
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30 214 fructose and glucose, which content increased during the initial phase of the  
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32 215 fermentation. In the exponential growth phase of the yeasts, glucose was the first sugar  
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34 216 to be fermented followed by fructose which degradation began in the stationary phase.  
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36 217 The conversion of sucrose, glucose and fructose to ethanol during the fermentation  
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38 218 process is the primary activity of the fermentative yeasts and the ethanol produced,  
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40 219 showed in Fig. 2, is, probably, a contribution for the beginning of the decline phase. In  
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42 220 fact, the yeasts' growth inhibition can be attributed either to toxicity of fermentation  
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44 221 products including ethanol, sugar transport inefficiency or other nutritional limitations  
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46 222 (Bisson, 1999).

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48 223 Although yeasts were not yet identified, it could be observed that a succession of  
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50 224 species took place as the fermentation proceeded. The earlier fermentation samples  
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52 225 contained a rich mixture of species. As fermentation progressed, the species diversity

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3 226 was reduced considerably and *Saccharomyces cerevisiae* (preliminary results) became  
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6 227 the dominant yeast at the end. Yeasts successions are known in numerous fermentation  
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8 228 processes as the fermentation of grape juice into wine (Fleet *et al.*, 1984; Querol *et al.*,  
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10 229 1990), the fermentation of sugars to produce tequila (Lachance, 1995) and in the  
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12 230 production of rum (Fahrasmane and Ganou-Parfait, 1998).  
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## 232 CONCLUSION

233 The purpose of this paper is to contribute to the understanding of this traditional spirit,  
234 in order to help optimized and standardize the fermentation process and consequently to  
235 obtain a more uniform composition and quality of this spirit. From the results obtained  
236 in natural fermentation, with wild microbiota, of *Arbutus unedo* fruit it can be  
237 concluded that the sugars increased during the first two days and gradually decreased  
238 along the fermentation period. At the same time, the mineral content was increasing  
239 resulting in a final product, in the must, enriched in Ca, Mg, Na and K. Additionally  
240 there was an increasing in the content of ethanol resulting from the alcoholic  
241 fermentation of the sugars performed by the yeasts. Although yeasts were not yet  
242 identified, it could be observed that a succession of species took place as the  
243 fermentation proceeded. *Aguardente de medronho* is a traditionally made distillate,  
244 generally of good quality, produced on a small scale from local distillers of the south  
245 region of Portugal. Maintaining the good quality of the product could contribute to the  
246 preservation and valorization of traditional resources that are of great importance to  
247 prevent their disappearance. Moreover arbutus berry show to be an appreciable source  
248 of sucrose and minerals. Further studies are necessary to investigate and identify the  
249 microbiological quality.

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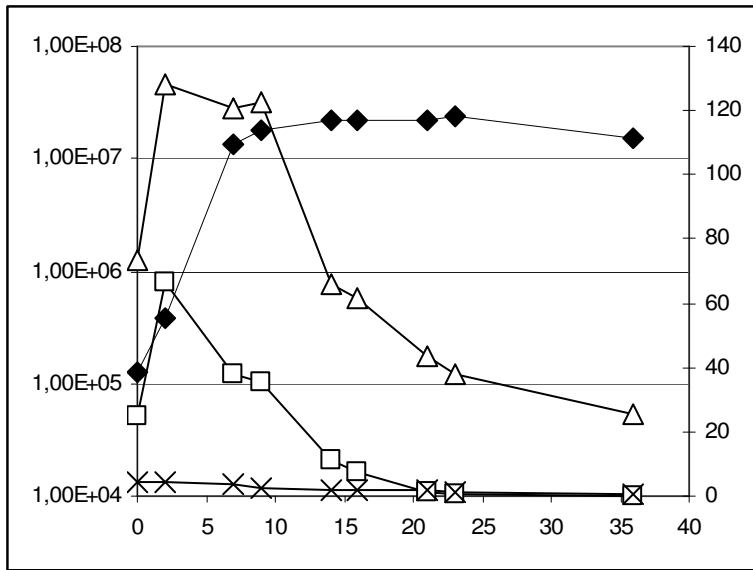
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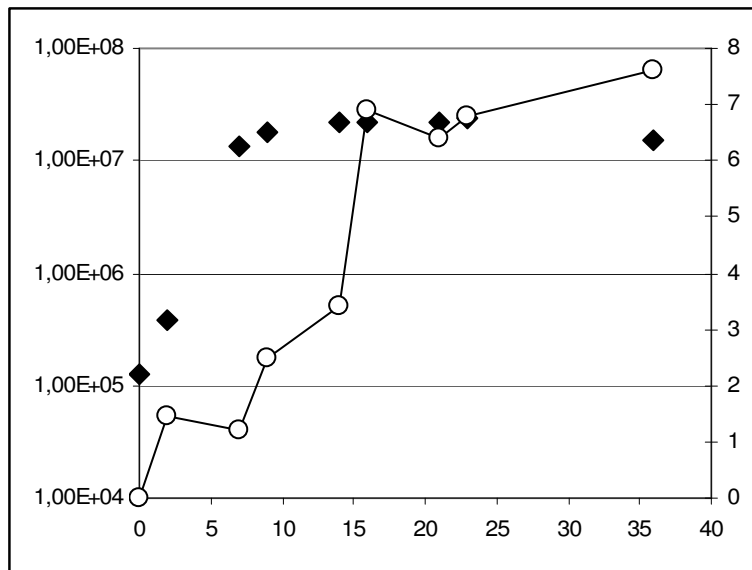
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**Fig. 1-** Changes in consumption of sugars (glucose (□), fructose (Δ), sucrose (×)) and yeast growth (◆) cycle during fermentation of *Arbutus unedo*' fruits.

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21 **Fig. 2.-** Production of ethanol (o) and yeast growth cycle(◆) during fermentation of

22 *Arbutus unedo*' fruits.

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**Table 1.** Changes in external and internal temperature of the bottle during the *Arbutus unedo*'s fruit fermentation.

Fermentation period (day)	External Temperature (°C)		Internal Temperature (°C)
	Máx.	Min.	
0	16.0	13.0	14.6
2	16.2	14.0	14.8
7	16.0	12.5	15.8
9	18.0	12.0	15.9
14	16.5	10.0	15.5
16	18.0	11.0	15.4
21	11.0	10.0	14.6
23	14.0	9.0	13.1
36	11.0	10.0	12.4

**Table 2.** Changes in pH and °Brix during the *Arbutus unedo*'s fruit fermentation.

<b>Fermentation period (day)</b>	<b>pH</b>	<b>° Brix</b>
0	3.28±0.1	23.50±0.1
2	3.28±0.1	23.00±0.1
7	4.03±0.1	22.00±0.1
9	4.43±0.1	-
14	4.62±0.1	17.00±0.1
16	4.94±0.1	15.80±0.1
21	4.74±0.1	14.00±0.1
23	5.06±0.1	13.50±0.1
36	4.82±0.1	12.00±0.1

**Table 3.** Mineral composition (mg/100g dry matter of seeds,  $n = 3$ ) during the *Arbutus unedo*'s fruit fermentation.

<b>Fermentation period (day)</b>	<b>Ca</b>	<b>Mg</b>	<b>Na</b>	<b>K</b>
0	1.665 <sup>a</sup> ±0.09	0.327 <sup>a</sup> ±0.091	1.021 <sup>a</sup> ±0.00	1.241 <sup>a</sup> ±0.01
2	1.629 <sup>a</sup> ±0.07	0.443 <sup>a</sup> ±0.01	1.021 <sup>a</sup> ±0.01	1.080 <sup>a</sup> ±0.00
7	1.933 <sup>a</sup> ±0.05	0.461 <sup>a</sup> ±0.01	1.153 <sup>a</sup> ±0.01	1.206 <sup>a</sup> ±0.00
9	2.258 <sup>b</sup> ±0.14	0.554 <sup>a</sup> ±0.01	1.123 <sup>a</sup> ±0.01	1.595 <sup>a</sup> ±0.00
14	1.905 <sup>b</sup> ±0.06	0.415 <sup>a</sup> ±0.02	0.923 <sup>a</sup> ±0.02	1.242 <sup>a</sup> ±0.01
16	2.163 <sup>b</sup> ±0.07	0.542 <sup>b</sup> ±0.02	1.059 <sup>a</sup> ±0.04	1.407 <sup>a</sup> ±0.02
21	2.606 <sup>c</sup> ±0.06	0.678 <sup>b</sup> ±0.02	1.257 <sup>a</sup> ±0.03	1.966 <sup>b</sup> ±0.02
23	3.368 <sup>c</sup> ±0.10	0.781 <sup>b</sup> ±0.01	1.209 <sup>a</sup> ±0.03	2.051 <sup>b</sup> ±0.01
36	3.036 <sup>c</sup> ±0.19	0.720 <sup>b</sup> ±0.03	1.138 <sup>a</sup> ±0.03	1.783 <sup>b</sup> ±0.01

Means ( $\pm$  s.d.) in the same column with different letters are significantly ( $P < 0.05$ ) different.